Separation of Protein Using ElectroMembrane: Study of Hydraulic-Electric Membrane Process to Purify Protein

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The purification of protein in this presented work is to investigate in the membrane process under the influence of an electric field. This paper presents with example from membrane process showing how the filtration time is reduced by the use of an electric field. The transmembrane pressure (TMP) was reduced by 20 % as the electric field increased. The concentration of protein in the membrane process in the presence of an electric field was reduced by over 300 % in comparison with the membrane process without an electric field. The hydraulic electrofiltration provides an another substitute to the crossflow filtration in the purification of protein.

Results

Theoretical Background

 $\begin{array}{l} AmA+CO_2\leftrightarrow AmACOO^{-}+AmAH^+\\ CO_2+OH^{-}\rightarrow HCO_3^{-}\\ RNHCOO^{-}+H_2O\leftrightarrow NH_2+HCO_3^{-}\\ R\cdot NH_2COONH_4+H_2O\rightarrow R\cdot (NH_4)\ _2CO_3\\ (Amine\ reaction\ with\ Aminoacid\ (Am:\ Amino\ acid) \end{array}$

Method

- Stirred Cell Reactor for Absorption Experiment -





- Membrane Apparatus for Protein Purification -





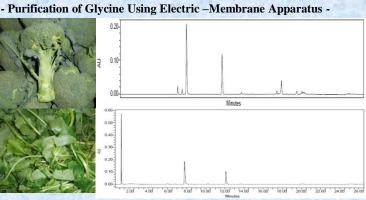




Conclusions

1. As the conc. of MEA absorbents increases by 6, 12, 18 wt%, the absorption rates were obtained as 3.0×10^{-6} , 6.0×10^{-6} , 7.8×10^{-6} mol/ cm²s.

2. But the same experiments were performed at the plant extracts, the absorption rate were obtained as 5.4×10^{-6} , 8.8×10^{-6} , $12.1 \times 10^{-6} \text{ mol/cm}^3$ s.



- Exp. Results of Enhancement of Aminoacid Conc.-

0.1N NaOH + Spinach	k _L (cm/s)		k (s ⁻¹)			D (cm²/s)	На
NaOH Only	4.38×10 ⁻³		6.08×10 ⁻³			1.13×1 0 ⁻⁷	5.98×1 0 ⁻³
1% Spinach	4.54×10 ⁻³		7.35×10 ⁻³		10 mm	1.14×1 0 ⁻⁷	6.38×1 0 ⁻³
5% Spinach	4.79×10 ⁻³		8.15×10 ⁻³			1.31×1 0 ⁻⁷	6.82×1 0 ⁻³
10% Spinach	5.19×10 ⁻³		9.96×10 ⁻³		NO MARK	1.63×1 0 ⁻⁷	7.36×1 0 ⁻³
Impact category		Unit		NaOH		MEA	Plant (spinach)
Global warming		kg CO ₂ eq		1.18		3.47	-1.52

Impact category	Umt	NaOH	MEA	(spinach)
Global warming (GWP 100)			3.47	-1.52
Ozone layer depletion (ODP)	kg CFC-11 eq	2.16E-7	2.96E-7	6.9E-9
Photochemical oxidation	kg C ₂ H ₄	0.0024	0.00179	5.11E-5
Acidification	kg SO ₂ eq	0.0156	0.0134	0.00634
Eutrophication	kg PO ₄ eq	0.000961	0.0036	0.00163
Non renewable, fossil	MJ eq	17.9	87	0.839